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# Ethyl 2-methyl-6-(propan-2-ylamino)-4-sulfanylidene-3*H*,11*H*-pyrimido[1,6-*c*]-quinazoline-1-carboxylate

**Hong-Xia Li,[a] Yu-Su Song,[a] Yong-nian Qu,[b] Jiang-Bing Lu[b] and Hong-Mei Wang[b]***

[a]College of Science, Naval University of Engineering, Wuhan 430033, People's Republic of China, and [b]Institute of Medicinal Chemistry, Hubei University of Medicine, Shi Yan 442000, People's Republic of China
Correspondence e-mail: meirwang@126.com

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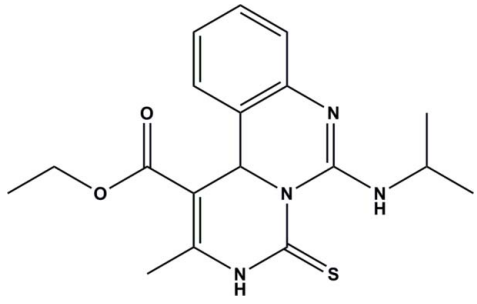
Key indicators: single-crystal X-ray study; $T$ = 298 K; mean $\sigma$(C–C) = 0.003 Å; $R$ factor = 0.050; $wR$ factor = 0.139; data-to-parameter ratio = 16.3.

The title compound, $C_{18}H_{22}N_4O_2S$, contains a substituted pyrimidine ring fused to both a benzene ring and a substituted thioxopyrimidine ring. The pyrimidine and thioxopyrimidine rings adopt distorted chair conformations. In the crystal, adjacent molecules are linked by pairs of N—H···S and N—H···O hydrogen bonds to generate centrosymmetric $R_2^2(8)$ and $R_2^2(16)$ loops, respectively. This combination leads to [100] chains of molecules.

## Related literature

For further synthetic details, see: Li *et al.* (2007, 2008); Huang *et al.* (2009); Zeng *et al.* (2010). For a related structure, see: Li *et al.* (2010). For ring conformations, see: Cremer & Pople (1975).




## Experimental

### Crystal data

$C_{18}H_{22}N_4O_2S$
$M_r$ = 358.46
Monoclinic, $P2_1/n$
$a$ = 9.4128 (3) Å
$b$ = 10.5636 (5) Å
$c$ = 19.2052 (6) Å
$\beta$ = 102.347 (1)°
$V$ = 1865.46 (12) Å$^3$
$Z$ = 4
Mo $K\alpha$ radiation
$\mu$ = 0.19 mm$^{-1}$
$T$ = 298 K
0.30 × 0.20 × 0.20 mm

### Data collection

Bruker SMART CCD diffractometer
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
$T_{min}$ = 0.955, $T_{max}$ = 0.962
12093 measured reflections
3857 independent reflections
3014 reflections with $I > 2\sigma(I)$
$R_{int}$ = 0.076

### Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.050
$wR(F^2)$ = 0.139
$S$ = 1.03
3857 reflections
236 parameters
2 restraints
H atoms treated by a mixture of independent and constrained refinement
$\Delta\rho_{max}$ = 0.34 e Å$^{-3}$
$\Delta\rho_{min}$ = −0.27 e Å$^{-3}$

**Table 1**
Hydrogen-bond geometry (Å, °).

| $D$—H···$A$ | $D$—H | H···$A$ | $D$···$A$ | $D$—H···$A$ |
|---|---|---|---|---|
| N2—H2$A$···O2$^{i}$ | 0.86 (1) | 2.26 (1) | 3.102 (2) | 165 (2) |
| N4—H4$A$···S1$^{ii}$ | 0.87 (1) | 2.42 (1) | 3.2804 (15) | 172 (2) |

Symmetry codes: (i) $-x, -y+1, -z+2$; (ii) $-x+1, -y+1, -z+2$.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT* (Bruker, 2001); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXTL* (Sheldrick, 2008).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB6711).

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# supplementary materials

*Acta Cryst.* (2012). E**68**, o1821 [doi:10.1107/S1600536812019952]

## Ethyl 2-methyl-6-(propan-2-ylamino)-4-sulfanylidene-3*H*,11*H*-pyrimido[1,6-*c*]quinazoline-1-carboxylate

**Hong-Xia Li, Yu-Su Song, Yong-nian Qu, Jiang-Bing Lu and Hong-Mei Wang**

### Comment

In recent years, we have been engaged in the preparation of heterocyclic derivatives using the aza-Wittig reaction (Li *et al.* 2007, 2008; Huang *et al.* 2009; Li *et al.* 2010; Zeng *et al.*, 2010). We present here the crystal structure of the title compound (Fig. 1). The molecule contains a substituted pyrimidine ring fused to a benzene ring and a substituted thioxo-pyrimidine ring. The centre ring of pyrimidine moiety adopts a distored chair conformation [$\varphi$ = 208.8 (2)° and $\theta$ = 68.85 (19)°, Puckering Amplitude = 0.5502 (17)Å], and the substituted thioxopyrimidine ring also show a distorted chair form [$\varphi$ = 214.6 (4)° and $\theta$ = 113.9 (3)°, Puckering Amplitude = 0.3827 (17)Å] (Cremer & Pople, 1975). In the crystal, there are N—H···O and N—H···S (Table 1) hydrogen bonds.

### Experimental

To a solution of iminophosphorane prepared according to Li *et al.* (2010) (0.55 g, 1 mmol) in $CH_3CN$ (10 mL) was added phenylisocyanate (0.12 g, 1 mmol) under nitrogen at room temperature. After stirred for 2 h at room temperature, $K_2CO_3$ (0.014 g,0.1 mmol) was added and the mixture was stirred for 1 h. The solvent was removed off under reduced pressure and the residue was recrystallized from methylene dichloride and ethanol to give the title compound (I) in yield of 85% (m.p. 490 K). Colourless blocks were obtained from a dichloromethane solution at room temperature.

### Refinement

The H atoms attached to atoms N2 and N4 was located in a difference Fourier map and allowed to ride on their parent atom with a restraint of N—H = 0.86 Å. Other H atoms were placed at calculated positions and treated as riding atoms, with C—H = 0.96–0.97 Å, and $U_{iso}$(H) = 1.2 $U_{eq}$(C) or 1.5 $U_{eq}$(C) for methyl H atoms.

### Computing details

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT* (Bruker, 2001); data reduction: *SAINT* (Bruker, 2001); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXTL* (Sheldrick, 2008).

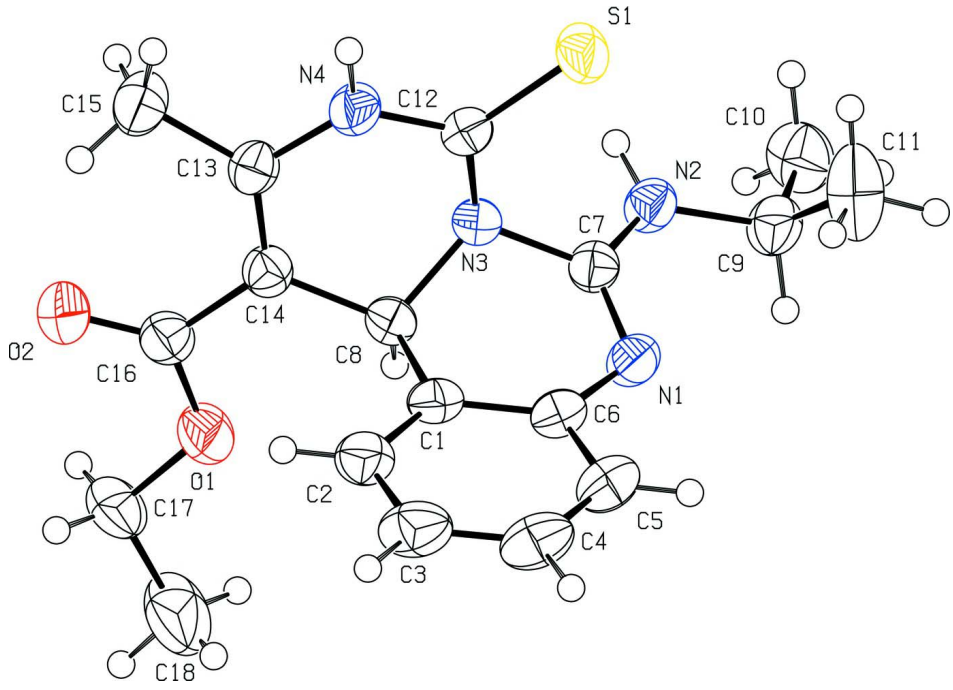


**Figure 1**

View of the molecule showing displacement ellipsoids drawn at the 50% probability level.

## Ethyl 2-methyl-6-(propan-2-ylamino)-4-sulfanylidene-3*H*,11*H*- pyrimido[1,6-*c*]quinazoline-1-carboxylate

*Crystal data*

$C_{18}H_{22}N_4O_2S$
$M_r = 358.46$
Monoclinic, $P2_1/n$
Hall symbol: -P 2yn
$a = 9.4128$ (3) Å
$b = 10.5636$ (5) Å
$c = 19.2052$ (6) Å
$\beta = 102.347$ (1)°
$V = 1865.46$ (12) Å$^3$
$Z = 4$

$F(000) = 760.0$
$D_x = 1.288$ Mg m$^{-3}$
Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Cell parameters from 4329 reflections
$\theta = 2.2$–$27.7°$
$\mu = 0.19$ mm$^{-1}$
$T = 298$ K
Block, colorless
0.30 × 0.20 × 0.20 mm

*Data collection*

Bruker SMART CCD diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
$\varphi$ and $\omega$ scans
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
$T_{min} = 0.955$, $T_{max} = 0.962$

12093 measured reflections
3857 independent reflections
3014 reflections with $I > 2\sigma(I)$
$R_{int} = 0.076$
$\theta_{max} = 26.5°$, $\theta_{min} = 2.2°$
$h = -11 \rightarrow 11$
$k = -13 \rightarrow 10$
$l = -24 \rightarrow 24$

*Refinement*

Refinement on $F^2$
Least-squares matrix: full
$R[F^2 > 2\sigma(F^2)] = 0.050$
$wR(F^2) = 0.139$
$S = 1.03$
3857 reflections
236 parameters
2 restraints
Primary atom site location: structure-invariant direct methods
Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H atoms treated by a mixture of independent and constrained refinement
$w = 1/[\sigma^2(F_o^2) + (0.0772P)^2 + 0.1069P]$ where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{max} = 0.025$
$\Delta\rho_{max} = 0.34$ e Å$^{-3}$
$\Delta\rho_{min} = -0.27$ e Å$^{-3}$

*Special details*

**Geometry**. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

**Refinement**. Refinement of $F^2$ against ALL reflections. The weighted R-factor wR and goodness of fit S are based on $F^2$, conventional R-factors R are based on F, with F set to zero for negative $F^2$. The threshold expression of $F^2 > 2$sigma($F^2$) is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on $F^2$ are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)*

| | *x* | *y* | *z* | $U_{iso}$*/$U_{eq}$ |
|---|---|---|---|---|
| C1 | 0.09011 (18) | 0.72990 (17) | 0.84262 (9) | 0.0364 (4) |
| C2 | 0.0719 (2) | 0.67844 (18) | 0.77505 (10) | 0.0462 (5) |
| H2 | 0.0294 | 0.5991 | 0.7656 | 0.055* |
| C3 | 0.1170 (3) | 0.7450 (2) | 0.72128 (10) | 0.0570 (6) |
| H3 | 0.1042 | 0.7107 | 0.6758 | 0.068* |
| C4 | 0.1809 (3) | 0.8623 (2) | 0.73568 (11) | 0.0558 (6) |
| H4 | 0.2120 | 0.9067 | 0.6998 | 0.067* |
| C5 | 0.1991 (2) | 0.91431 (19) | 0.80297 (10) | 0.0494 (5) |
| H5 | 0.2422 | 0.9935 | 0.8120 | 0.059* |
| C6 | 0.1535 (2) | 0.84940 (17) | 0.85729 (9) | 0.0380 (4) |
| C7 | 0.17531 (19) | 0.83292 (16) | 0.97816 (9) | 0.0348 (4) |
| C8 | 0.04635 (19) | 0.66650 (16) | 0.90571 (9) | 0.0345 (4) |
| H8 | −0.0437 | 0.7061 | 0.9126 | 0.041* |
| C9 | 0.2171 (2) | 1.00033 (18) | 1.06939 (10) | 0.0466 (5) |
| H9 | 0.1470 | 1.0559 | 1.0387 | 0.056* |
| C10 | 0.1933 (3) | 1.0103 (2) | 1.14433 (12) | 0.0612 (6) |
| H10A | 0.0963 | 0.9833 | 1.1453 | 0.092* |
| H10B | 0.2063 | 1.0965 | 1.1601 | 0.092* |
| H10C | 0.2621 | 0.9573 | 1.1753 | 0.092* |
| C11 | 0.3679 (3) | 1.0396 (2) | 1.06328 (15) | 0.0795 (8) |
| H11A | 0.4385 | 0.9897 | 1.0953 | 0.119* |
| H11B | 0.3823 | 1.1275 | 1.0754 | 0.119* |
| H11C | 0.3791 | 1.0265 | 1.0153 | 0.119* |
| C12 | 0.28831 (19) | 0.62791 (16) | 0.97962 (9) | 0.0343 (4) |
| C13 | 0.13802 (19) | 0.44969 (16) | 0.92654 (9) | 0.0363 (4) |

| | | | | |
|---|---|---|---|---|
| C14 | 0.02360 (19) | 0.52521 (16) | 0.90196 (9) | 0.0355 (4) |
| C15 | 0.1435 (2) | 0.30767 (17) | 0.92606 (12) | 0.0502 (5) |
| H15A | 0.1103 | 0.2753 | 0.9665 | 0.075* |
| H15B | 0.2417 | 0.2804 | 0.9284 | 0.075* |
| H15C | 0.0821 | 0.2765 | 0.8830 | 0.075* |
| C16 | −0.1233 (2) | 0.47530 (18) | 0.87161 (9) | 0.0404 (4) |
| C17 | −0.3692 (2) | 0.5353 (2) | 0.82596 (12) | 0.0552 (5) |
| H17A | −0.4100 | 0.4905 | 0.8612 | 0.066* |
| H17B | −0.3745 | 0.4806 | 0.7849 | 0.066* |
| C18 | −0.4504 (3) | 0.6539 (3) | 0.8048 (2) | 0.0979 (11) |
| H18A | −0.4431 | 0.7077 | 0.8457 | 0.147* |
| H18B | −0.5508 | 0.6345 | 0.7856 | 0.147* |
| H18C | −0.4101 | 0.6966 | 0.7693 | 0.147* |
| S1 | 0.44958 (5) | 0.68615 (5) | 1.01900 (3) | 0.04898 (19) |
| N1 | 0.16807 (17) | 0.90727 (14) | 0.92474 (8) | 0.0399 (4) |
| N2 | 0.18611 (17) | 0.86974 (14) | 1.04561 (8) | 0.0387 (4) |
| H2A | 0.191 (2) | 0.8115 (14) | 1.0774 (8) | 0.046* |
| N3 | 0.16359 (15) | 0.69732 (13) | 0.96840 (7) | 0.0326 (3) |
| N4 | 0.27056 (16) | 0.50620 (14) | 0.95704 (8) | 0.0393 (4) |
| H4A | 0.3503 (15) | 0.4619 (16) | 0.9646 (10) | 0.047* |
| O1 | −0.21898 (14) | 0.57011 (13) | 0.85569 (8) | 0.0525 (4) |
| O2 | −0.15658 (16) | 0.36556 (14) | 0.86187 (8) | 0.0569 (4) |

*Atomic displacement parameters (Å²)*

| | $U^{11}$ | $U^{22}$ | $U^{33}$ | $U^{12}$ | $U^{13}$ | $U^{23}$ |
|---|---|---|---|---|---|---|
| C1 | 0.0333 (9) | 0.0385 (10) | 0.0363 (9) | 0.0059 (8) | 0.0045 (7) | 0.0066 (7) |
| C2 | 0.0516 (12) | 0.0455 (11) | 0.0392 (10) | 0.0030 (9) | 0.0045 (9) | 0.0016 (8) |
| C3 | 0.0744 (15) | 0.0612 (14) | 0.0352 (10) | 0.0142 (12) | 0.0113 (10) | 0.0055 (9) |
| C4 | 0.0695 (15) | 0.0559 (13) | 0.0460 (11) | 0.0109 (11) | 0.0215 (10) | 0.0180 (10) |
| C5 | 0.0567 (13) | 0.0425 (11) | 0.0509 (11) | 0.0035 (9) | 0.0163 (10) | 0.0131 (9) |
| C6 | 0.0387 (10) | 0.0348 (10) | 0.0401 (9) | 0.0068 (8) | 0.0077 (8) | 0.0051 (7) |
| C7 | 0.0306 (9) | 0.0329 (9) | 0.0409 (9) | 0.0019 (7) | 0.0077 (7) | −0.0004 (7) |
| C8 | 0.0298 (8) | 0.0366 (10) | 0.0357 (8) | 0.0006 (7) | 0.0041 (7) | 0.0004 (7) |
| C9 | 0.0554 (12) | 0.0361 (10) | 0.0481 (11) | 0.0040 (9) | 0.0107 (9) | −0.0051 (8) |
| C10 | 0.0690 (15) | 0.0603 (14) | 0.0544 (12) | 0.0072 (12) | 0.0140 (11) | −0.0168 (10) |
| C11 | 0.0865 (19) | 0.0674 (17) | 0.0933 (19) | −0.0312 (14) | 0.0387 (16) | −0.0277 (14) |
| C12 | 0.0369 (9) | 0.0324 (9) | 0.0338 (8) | 0.0020 (7) | 0.0077 (7) | 0.0034 (7) |
| C13 | 0.0379 (10) | 0.0346 (9) | 0.0379 (9) | −0.0025 (7) | 0.0113 (7) | 0.0015 (7) |
| C14 | 0.0362 (9) | 0.0358 (10) | 0.0348 (9) | −0.0028 (7) | 0.0085 (7) | 0.0006 (7) |
| C15 | 0.0483 (12) | 0.0375 (11) | 0.0662 (13) | −0.0022 (9) | 0.0150 (10) | 0.0009 (9) |
| C16 | 0.0414 (10) | 0.0440 (11) | 0.0356 (9) | −0.0047 (9) | 0.0074 (8) | 0.0039 (8) |
| C17 | 0.0373 (11) | 0.0642 (14) | 0.0587 (12) | −0.0093 (10) | −0.0021 (9) | −0.0010 (10) |
| C18 | 0.0541 (16) | 0.0719 (18) | 0.148 (3) | 0.0067 (13) | −0.0231 (17) | −0.0301 (18) |
| S1 | 0.0332 (3) | 0.0401 (3) | 0.0684 (4) | 0.00183 (19) | −0.0007 (2) | −0.0082 (2) |
| N1 | 0.0450 (9) | 0.0325 (8) | 0.0420 (8) | 0.0024 (7) | 0.0088 (7) | 0.0039 (6) |
| N2 | 0.0451 (9) | 0.0327 (8) | 0.0389 (8) | 0.0017 (7) | 0.0104 (7) | −0.0004 (6) |
| N3 | 0.0310 (7) | 0.0323 (8) | 0.0335 (7) | 0.0012 (6) | 0.0043 (6) | 0.0010 (6) |
| N4 | 0.0338 (8) | 0.0320 (8) | 0.0507 (9) | 0.0033 (6) | 0.0057 (7) | 0.0006 (7) |
| O1 | 0.0345 (7) | 0.0504 (9) | 0.0675 (9) | −0.0043 (6) | −0.0002 (6) | −0.0017 (7) |

| | | | | | | |
|---|---|---|---|---|---|---|
| O2 | 0.0522 (9) | 0.0447 (9) | 0.0673 (9) | −0.0126 (7) | −0.0016 (7) | 0.0034 (7) |

*Geometric parameters (Å, º)*

| | | | |
|---|---|---|---|
| C1—C2 | 1.384 (2) | C11—H11A | 0.9600 |
| C1—C6 | 1.399 (3) | C11—H11B | 0.9600 |
| C1—C8 | 1.516 (2) | C11—H11C | 0.9600 |
| C2—C3 | 1.389 (3) | C12—N4 | 1.356 (2) |
| C2—H2 | 0.9300 | C12—N3 | 1.362 (2) |
| C3—C4 | 1.380 (3) | C12—S1 | 1.6625 (18) |
| C3—H3 | 0.9300 | C13—C14 | 1.342 (2) |
| C4—C5 | 1.381 (3) | C13—N4 | 1.394 (2) |
| C4—H4 | 0.9300 | C13—C15 | 1.501 (2) |
| C5—C6 | 1.390 (3) | C14—C16 | 1.478 (2) |
| C5—H5 | 0.9300 | C15—H15A | 0.9600 |
| C6—N1 | 1.412 (2) | C15—H15B | 0.9600 |
| C7—N1 | 1.282 (2) | C15—H15C | 0.9600 |
| C7—N2 | 1.335 (2) | C16—O2 | 1.205 (2) |
| C7—N3 | 1.446 (2) | C16—O1 | 1.338 (2) |
| C8—N3 | 1.485 (2) | C17—O1 | 1.454 (2) |
| C8—C14 | 1.507 (2) | C17—C18 | 1.479 (3) |
| C8—H8 | 0.9800 | C17—H17A | 0.9700 |
| C9—N2 | 1.463 (2) | C17—H17B | 0.9700 |
| C9—C10 | 1.507 (3) | C18—H18A | 0.9600 |
| C9—C11 | 1.507 (3) | C18—H18B | 0.9600 |
| C9—H9 | 0.9800 | C18—H18C | 0.9600 |
| C10—H10A | 0.9600 | N2—H2A | 0.861 (9) |
| C10—H10B | 0.9600 | N4—H4A | 0.870 (9) |
| C10—H10C | 0.9600 | | |

| | | | |
|---|---|---|---|
| C2—C1—C6 | 120.35 (16) | H11A—C11—H11C | 109.5 |
| C2—C1—C8 | 125.15 (16) | H11B—C11—H11C | 109.5 |
| C6—C1—C8 | 114.50 (15) | N4—C12—N3 | 114.64 (15) |
| C1—C2—C3 | 120.16 (19) | N4—C12—S1 | 122.31 (13) |
| C1—C2—H2 | 119.9 | N3—C12—S1 | 123.04 (13) |
| C3—C2—H2 | 119.9 | C14—C13—N4 | 118.16 (16) |
| C4—C3—C2 | 119.69 (19) | C14—C13—C15 | 128.17 (17) |
| C4—C3—H3 | 120.2 | N4—C13—C15 | 113.67 (16) |
| C2—C3—H3 | 120.2 | C13—C14—C16 | 122.63 (17) |
| C3—C4—C5 | 120.42 (19) | C13—C14—C8 | 118.46 (16) |
| C3—C4—H4 | 119.8 | C16—C14—C8 | 118.91 (15) |
| C5—C4—H4 | 119.8 | C13—C15—H15A | 109.5 |
| C4—C5—C6 | 120.66 (19) | C13—C15—H15B | 109.5 |
| C4—C5—H5 | 119.7 | H15A—C15—H15B | 109.5 |
| C6—C5—H5 | 119.7 | C13—C15—H15C | 109.5 |
| C5—C6—C1 | 118.72 (17) | H15A—C15—H15C | 109.5 |
| C5—C6—N1 | 119.32 (17) | H15B—C15—H15C | 109.5 |
| C1—C6—N1 | 121.92 (15) | O2—C16—O1 | 122.99 (17) |
| N1—C7—N2 | 125.29 (16) | O2—C16—C14 | 126.50 (18) |
| N1—C7—N3 | 120.91 (15) | O1—C16—C14 | 110.52 (16) |

| | | | |
|---|---|---|---|
| N2—C7—N3 | 113.72 (15) | O1—C17—C18 | 107.19 (18) |
| N3—C8—C14 | 109.22 (13) | O1—C17—H17A | 110.3 |
| N3—C8—C1 | 105.62 (13) | C18—C17—H17A | 110.3 |
| C14—C8—C1 | 117.33 (14) | O1—C17—H17B | 110.3 |
| N3—C8—H8 | 108.1 | C18—C17—H17B | 110.3 |
| C14—C8—H8 | 108.1 | H17A—C17—H17B | 108.5 |
| C1—C8—H8 | 108.1 | C17—C18—H18A | 109.5 |
| N2—C9—C10 | 107.66 (17) | C17—C18—H18B | 109.5 |
| N2—C9—C11 | 111.32 (17) | H18A—C18—H18B | 109.5 |
| C10—C9—C11 | 112.96 (19) | C17—C18—H18C | 109.5 |
| N2—C9—H9 | 108.3 | H18A—C18—H18C | 109.5 |
| C10—C9—H9 | 108.3 | H18B—C18—H18C | 109.5 |
| C11—C9—H9 | 108.3 | C7—N1—C6 | 116.54 (15) |
| C9—C10—H10A | 109.5 | C7—N2—C9 | 123.11 (15) |
| C9—C10—H10B | 109.5 | C7—N2—H2A | 117.5 (13) |
| H10A—C10—H10B | 109.5 | C9—N2—H2A | 118.4 (13) |
| C9—C10—H10C | 109.5 | C12—N3—C7 | 118.27 (14) |
| H10A—C10—H10C | 109.5 | C12—N3—C8 | 118.45 (14) |
| H10B—C10—H10C | 109.5 | C7—N3—C8 | 110.19 (13) |
| C9—C11—H11A | 109.5 | C12—N4—C13 | 125.35 (15) |
| C9—C11—H11B | 109.5 | C12—N4—H4A | 114.4 (14) |
| H11A—C11—H11B | 109.5 | C13—N4—H4A | 120.2 (14) |
| C9—C11—H11C | 109.5 | C16—O1—C17 | 116.80 (15) |

*Hydrogen-bond geometry (Å, º)*

| *D*—H···*A* | *D*—H | H···*A* | *D*···*A* | *D*—H···*A* |
|---|---|---|---|---|
| N2—H2*A*···O2$^{i}$ | 0.86 (1) | 2.26 (1) | 3.102 (2) | 165 (2) |
| N4—H4*A*···S1$^{ii}$ | 0.87 (1) | 2.42 (1) | 3.2804 (15) | 172 (2) |

Symmetry codes: (i) −*x*, −*y*+1, −*z*+2; (ii) −*x*+1, −*y*+1, −*z*+2.